Sintering Trials of Analogues of Americium Oxides for Radioisotope Power Systems

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Abstract

European Space Agency radioisotope power systems will use americium oxide as the heat source in pellet or disc form. The oxide form is yet to be decided. Sintering trials with CeO_2 and Nd_2O_3 as analogues for AmO_2 and Am_2O_3 were conducted. Spark plasma sintering (SPS) and cold-press-and-sinter methods were compared. Different sintering parameters and particle characteristics were investigated with commercial and synthesised powders. The synthesised powders contained lath-shaped particles and batches with different particle sizes and specific surface areas were made and sintered. This is the first study in the public literature to report the sintering of lath-shaped CeO₂.

The targeted density range of 85-90 % was met using both techniques. No ball-milling was required. Cold-pressing-and-sintering CeO₂ produced intact discs. Large cracking was prevalent in the SPS discs. Some powders pressed more successfully than others. Powder morphology had a significant effect on the result but it was not possible to fully quantify the effects in this study. The cold-pressed-andsintered CeO₂ discs had comparable Vickers hardness values to a nuclear ceramic (UO₂). The hardness values were greater than the spark plasma sintered CeO₂ sample. Efforts to SPS near-net shaped pellets using CeO₂ and Nd₂O₃ are reported. A follow on investigation was conducted to assess how the 85-90 % T.D. target could be achieved. The aspect ratio impacts the sintering parameters and behaviour. The Vickers hardness of Nd₂O₃ is reported for the first time and compared to the results of sintered CeO₂.

1 Introduction

Radioisotope power systems (RPS) enable deep space exploration as well as the ability to probe some of the more challenging environments on the surfaces of solar system bodies. Unlike solar power systems, RPSs are independent of the local solar intensity. The versatility of radioisotope thermoelectric generators (RTGs) and radioisotope heater units (RHUs) has enabled a range of space missions since the 1960s that would not have been possible with solar power [1-3].

The European Space Agency (ESA) started a research and development programme in 2008 to develop a European RPS [4-7]. Americium-241 (²⁴¹Am) was identified as a viable and affordable fuel for Europe [4]. It was selected as the radioisotope fuel despite its reduced specific thermal power compared to ²³⁸Pu [8]. The ESA programme has since progressed to the production and processing

of americium oxide, and the development of different RPS technologies including RTGs [5], RHUs and Stirling generators [4].

Different forms of americium oxide are being investigated to select a preferred fuel form. These include oxides ranging from Am_2O_3 to AmO_2 . The material will need to be in a form that can be integrated easily into an RPS system. Historically, the RPS systems built by the United States have used consolidated PuO_2 with various geometries including spheres, discs and pellets [1].

The overarching aim of this study was to investigate and demonstrate the sintering of suitable surrogate oxides for americium oxides whilst comparing the performance of the cold-press-and-sinter and spark plasma sintering (SPS) approaches. The requirement was for a reproducible process to produce an intact cylindrical pellet of ~13 mm diameter by ~16 mm long. It was sized assuming a requirement for an RHU generating several watts of thermal power (~3 W_{th}). Two main driving constraints were considered. Firstly that the relative density of the material should be 85-90% of theoretical density to allow helium escape during operation. Secondly, for radiological protection reasons, the powder used for the sintering should not be subject to any ball-milling prior to sintering. This is both to reduce respirable fines and equipment contamination [9].

In conventional cold-press-and-sinter, powders are compacted under pressure (typically by using a steel die) at room temperature to improve their packing density to form 'green' bodies [10]. The green discs or pellets are then heated in a furnace according to a specified time-temperature cycle to sinter the compact. The sintering dwell time for cold-pressed-and-sintered bodies can be several hours. In spark plasma sintering, the heat is generated by direct Joule heating of conductive dies and the powder if it is sufficiently electrically conductive. This allows high sintered densities to be achieved much more quickly and/or at lower temperatures than conventional methods. It also has the potential to manufacture near-net shaped pellets [11]. Graphite dies are often used and so SPS is described as providing a reducing sintering environment for sintering oxides [12]; this is a potential disadvantage compared to the conventional cold-press-and-sinter approach.

Non-radioactive surrogate materials provide early insights into how the fuel will behave without the challenges of working with active material [13]. Neodymium (III) oxide (Nd₂O₃), cerium (IV) oxide (CeO₂) and solid solution oxides of the two have been identified as appropriate surrogates for Am₂O₃, AmO₂ and sub-stoichiometric AmO₂, respectively, based on crystal phases and cationic radii, and on transition temperatures, melting points and thermal expansion coefficients [7, 13, 14]. The current ESA americium fuel chemical flow-sheet creates AmO₂ particles with lath-shaped morphology of tens of micrometres in size [15]. Production of surrogate CeO₂ material using a directly analogous oxalate precipitation and calcination process has also been demonstrated to produce micrometric lath-shaped CeO₂ particles [16].

Cold-press-and-sinter has been successfully demonstrated for nanometric CeO₂ powders [17-22]. Near full density was achieved in a number of these studies. Reduction of the ceria, resulting in oxygen outgassing, pore production and reduced relative densities of between 70-85 % was observed in some studies [17, 18, 23]. Wang [19] demonstrated the cold-press-and-sinter of CeO₂ at a larger powder size of 5 μ m; however, this was ball-milled before sintering. Roleček et al. [24] reported a comparison of cold-press-and-sintered milled and non-milled powder. The relative densities achieved were 72.2 % to 91.3 % when non-milled powder was used, and 81.4 % to 95.4 % when milled powder was used. Data for particle size and morphology of the powder (in particular whether the powder is lath-shaped) were not reported by Roleček et al. [24]. Cold-press-and-sinter of Nd₂O₃ has been reported [25] using powder with an average grain size of 4 μ m, achieving a relative density of 86-89 % but with cracking attributed to a hygroscopic reaction.

There has been limited research into the SPS of pure or doped CeO₂. Choi et al. [26, 27] investigated the SPS processing of pure nanometric CeO₂ and achieved greater than 97 % relative density. Roleček et al. [24] reported a comparative study on the SPS and cold-press-and-sinter of CeO₂. SPS allowed a much reduced processing time (5 min *cf* 1 hour) and pressure (50 MPa *cf* 300 MPa); however, there was XRD evidence of reduction to Ce₂O₃. Mori et al. [28] also reported difficulties in the densification of doped nano-CeO₂ processed using SPS due to reduction. The SPS of ball-milled Nd₂O₃ powder, which had a nominal particle of size of 4 μ m prior to milling, was demonstrated to achieve high relative density of 98.8 % [7].

One of the few mechanical properties reported for actinide oxides and their surrogates is Vickers hardness [7, 29, 30]. This standard technique can be conducted on small samples of material [31], and the data obtained allow an initial comparison with collated values of similar materials. Jahromi [22] measured the Vickers hardness of cold-press-and-sintered CeO₂ using a 0.25 N testing load to be between 4.6 GPa and 7.7 GPa depending on sintering temperature. No open literature reporting Vickers hardness for pure Nd₂O₃ has been found.

The objective of this study was therefore to investigate spark plasma sintering (SPS) and conventional cold-pressing-and-sintering of micrometric CeO_2 and Nd_2O_3 . Wherever practical, these studies would use lath-shaped micrometric CeO_2 produced directly from the wet chemical process that would be applicable to the eventual flight fuel form. In all cases, ball-milling of the powders would be avoided. The study is exploratory with no known public studies on the SPS of such large particulate CeO_2 or on the effect of the lath particle morphology on sintering behaviour.

The development of an SPS process using commercially available micrometric CeO₂ sintered in two stages to manage the reduction and outgassing is reported. These sintering parameters were then used to produce near-net shaped ceria pellets using lath-shaped material. Pellets were also produced using conventional cold-press-and-sinter from lath-shaped material of different particle sizes and specific surface areas. Finally, the SPS of commercially available Nd₂O₃ was performed to demonstrate the production of near-net shape pellets and for Vickers hardness measurement.

2 Method

2.1 Materials

Pale yellow commercial CeO₂ powder (99.9 % purity, Sigma Aldrich, UK) with a particle size <5 μ m was procured for use in the initial SPS studies. This approach was taken because the wet chemical process for synthesising CeO₂ produces small volumes of material and more material was needed for initial trials. The as-received particles are shown in Figure 1a.

Wet-chemically synthesised CeO₂ was prepared via an oxalate precipitation and calcination process. The former used 0.083 M cerium (III) nitrate hexahydrate in 0.2 M nitric acid and 0.68 M oxalic acid. The method is a continuous precipitation process [16] and is similar to a process previously described [13]. The oxalate precipitation temperature and calcination temperatures used are given in Table 1. This process yields lath-shaped particles as shown in Figure 1b. A Malvern Mastersizer 3000 was used to conduct the particle size analysis using a wet dispersion technique. It used low angle laser light scattering to obtain particle size distributions. A Micrometrics Tristar2 was used to measure the SSA of the powder samples using the Brunauer-Emmett-Teller (BET) method. Table 1 shows that the calcination temperature has a significant effect on specific surface area, with a lower calcination temperature resulting in a higher specific surface area. The impact of the precipitation and calcination conditions on particle size was more subtle as shown by Table 1. Commercially sourced neodymium(III) oxide powder (99.9 % purity, Alfa Aesar, UK) was procured for the SPS trials using the same product code as stated in a previous SPS study, which had a particle size of approximately 4 μ m [7].

Batch #	Oxalate Precipitation	Calcination Temp. (°C)	Particle Size (µm)				Specific Surface Area (m ² g ⁻¹)
	Temp. (°C)		d10 d50 d90		Volume Mean		
1	25	500	5	25	68	32	67
2	25	650	6	27	70	33	20
3	25	900	6	27	73	38	2
4	60	500	5	23	78	33	69
5	60	650	4	13	37	17	17
6	60	900	5	20	62	28	4

Table 1: Outlines the wet-chemical synthesis oxalate precipitation and calcination conditions used to make the different batches of lath-shaped CeO_2 , together with their sizes and specific surface areas.



Figure 1: SEM images of (a) the as received CeO_2 commercial material and (b) an example of synthesised lath-shaped CeO_2 .

2.2 Sintering Profiles and Methodology

2.2.1 Spark Plasma Sintering of Cerium (IV) Oxide

Initial trials investigated the heating rate, peak temperature, pressure, hold time and the cooling conditions for producing 20 mm diameter disc specimens using a single step sintering profile. The main parameters investigated were the variation of temperature between 1100 °C and 1500 °C and pressures of 50 MPa and 80 MPa. The pressure of gas evolved from the specimen during processing

was measured by the SPS furnace. A graphite die (Duragraph 20) was used with graphite paper (Papyex N998) to aid electrical contact.

Sintering profiles A and B, which are illustrated in Figure 2, were used to sinter 20 mm diameter discs (A and B) using commercial CeO₂ powder. The first sintering stage in each profile targeted 1050 °C and 1100 °C, respectively, to allow the material to reduce and lose oxygen at a lower relative density. The higher temperatures and pressures of the second stage of each profile were used to consolidate the reduced ceria. The discs were allowed to free cool.

The wet-chemically synthesised lath-shaped CeO₂ from batches 1 and 2 were sintered using sintering profile C, which is illustrated in Figure 2, to target the fabrication of cylindrical near-net shaped pellets. A graphite die (Duragraph 465) with an internal diameter of 13 mm was used with a target pellet length of 16 mm. The sintering profile was based on Condition B but featured a more gradual cooling profile (20 °C min⁻¹ from 1300 °C to 450 °C and then to room temperature with no pressure applied) and a lower pressure of 50 MPa. These changes were made with the aim of reducing the risks of failure of the graphite die and pellet cracking due to thermal stresses.



Figure 2: SPS Sinter profiles. a) Two-stage SPS Condition A used with commercial CeO_2 ; b) Two-stage SPS Condition B used with commercial CeO_2 and c) Two-stage SPS Condition C used with lath-shaped CeO_2 and commercial Nd_2O_3 to produce near-net shape pellets.

2.2.2 Cold-Press-and–Sinter of Cerium (IV) Oxide

Wet chemically synthesised lath-shaped CeO_2 from batches 1-6 (see Table 1) were pressed into discs using a 53 kN load and a die with inner dimension of 8 mm. No binder was used. It was not possible to produce intact green discs out of CeO_2 batches 4 and 6. Multiple attempts were made with batch 4. A programmable furnace with $MoSi_2$ u-pin elements (CM Inc., Bloomfield, NJ) was used to sinter the green bodies. The sintering profile used is shown schematically in Figure 3.



Figure 3: The sintering profile used to sinter the cold pressed lath-shaped CeO_2 discs in air.

2.2.3 Spark Plasma Sintering of Neodymium (III) Oxide

Sintering profile Condition C, which is depicted in Figure 2, was used to demonstrate whether a near-net shaped pellet could be fabricated from commercially procured Nd_2O_3 . Sample dimensions were the same as those for the CeO₂ pellets (section 2.2.1). Condition C was adapted by varying the temperature between 1000 °C and 1350 °C, whilst keeping the pressure at 50 MPa, to identify a suitable temperature to achieve the target relative density of 85-90 %. A further near-net shape pellet was then sintered using the temperature identified by this part of the study to investigate the effect of the aspect ratio on the sintering behaviour.

2.3 Characterisation

The relative density of the discs was calculated geometrically, rather than using an Archimedes method, in order to prevent potential degradation of the specimens due to immersion in any liquid. All CeO₂ materials were nominally assumed to have a theoretical density of 7.20 g cm⁻³. The theoretical density of Nd₂O₃ was taken to be 7.24 g cm⁻³. A combined uncertainty estimate on the relative density was produced for all specimens based on measurement precision. Specimens were stored in sealed vacuum bags in a dry cabinet for between one week and nine months between production, inspection and testing.

Microscope and visual inspection of the specimens was performed to investigate the microstructure and to confirm whether intact pellets could be produced for each material and sintering condition.

Vickers hardness tests were used to assess the mechanical properties of materials produced since this is one of the few mechanical parameters for which literature data for actinide and lanthanide oxides were available. Measurements were performed following ASTM standard C1327-08 [31] using a 300 g load and 15 s dwell time on ground and polished specimens cold mounted in epoxy resin at ambient conditions. Ten acceptable indents were used to calculate a mean and standard deviation of the hardness for each test case. The CeO₂ spark plasma sintered discs were mounted such that their fractured surfaces would be polished. The near-net shaped SPS pellet, cold-pressed-and-sintered CeO₂ discs and an Nd₂O₃ disc produced using SPS were mounted with their circular faces upwards. Vickers hardness tests were performed at two different institutions and by different operators. This meant they were potentially susceptible to subjective differences such as in judging acceptability of indentation morphology and minor differences in procedures such as repeat measurements of individual indentations. For example, scanning electron microscopy (SEM) was used to measure the indents in the CeO₂ SPS discs because material uplift made it difficult to obtain focussed optical images. These factors should be considered when comparing the results. However, in this case the objective is to generate indicative data only and so any procedural difference does not place major limitations on the study conclusions.

3 Results

3.1 Cerium (IV) Oxide

3.1.1 Behaviour of CeO₂ under Spark Plasma Sintering

The results of the one-stage initial SPS sintering trials using commercially sourced CeO_2 (§ 2.2.1) are shown in Table 2. The discs cracked on removal from the die. Visual observation of the discs revealed they were generally grey rather than pale yellow, which was the colour of the powder prior to sintering.

The results in Table 2 show that a small change in parameters can have a large effect on the relative density of the disc. Achieving the required relative density range of 85-90% can be readily achieved by selecting the appropriate conditions. Figure 4 illustrates the fractured surfaces of some of the produced discs. It qualitatively illustrates that an increase in temperature improved consolidation. The impact of increasing the temperature from 1200 °C to 1300 °C under an 80 MPa pressure for 3 min resulted in grain growth. Increasing the sintering temperature to 1500 °C resulted in a significant change in fracture surface microstructure. This is illustrated in Figure 5, which is at a higher magnification than Figure 4. The sintering is improved in the 1500 °C disc; however, its angular surfaces suggest grain pull-out and non-optimal consolidation. Figure 6 shows that outgassing was detected in one of the early trial discs as the temperature passed 1025°C.

Temperature (°C)	Pressure (MPa)	Time (min)	Mass (g)	Diameter (mm)	Thickness (mm)	Sintered Density (g cm ⁻³)	% Theoretical Density (T.D.)	SPS Machine Left to Free Cool at high temperature? (Yes/No)
1100	50	3	7.2	20	4.3	5.3	74 ± 1.4	No
1250	50	3	7.2	20	3.8	6.0	84 ± 1.7	Yes
1200	80	0	7.2	20	4.5	5.1	71 ± 1.3	No
1200	80	3	7.2	20	3.9	5.9	82 ± 1.6	No
1300	80	3	7.2	20	3.6	6.4	88 ± 1.9	No
1500	80	3	7.2	20	3.5	6.5	91 ± 2.0	No

Table 2: The parameters used for single-stage SPS using commercially sourced CeO₂.



Figure 4: The microstructures (5 μm scale) of some of the single-stage SPS ceria made under the following conditions: a) 1200 °C/80 MPa/3 min and b) 1300 °C/80 MPa/3 min using commercial CeO₂.



Figure 5: The microstructure of a single-stage SPS disc made at 1500 °C/80 MPa/3 min using commercial CeO₂.





The objective of using a two-stage SPS process with commercially sourced CeO₂ was to allow the material to outgas and reduce at a lower temperature and pressure and then to allow the ceria to sinter at a higher temperature and pressure in order to mitigate the cracking observed in the specimens manufactured in a single stage. The fractured surfaces of the two-stage SPS discs A and B (made under conditions A and B shown in Figure 2a and 2b, respectively), which were manufactured using commercial CeO₂, are illustrated in Figure 7. Disc A was intact and disc B was broken on removal from the die. A small change in sintering profile from condition A to B resulted in a large change in relative density and microstructure: discs A and B had relative densities of 79 ± 1.2 % and 88 ± 1.4 %, respectively, and disc B evidenced improved sintering when compared to disc A. Disc B also had small pores. The ground and polished fragments of disc A and disc B are shown in Figure 8. Disc A had a central lighter region and a darker border. It was not clear whether there was an evolution over time in the colour, and therefore in the oxygen to metal ratio, of this central region during the several months between when the disc was made, stored, and then ground and polished for analysis. To aid seeing this colour difference effect, the image contrast and brightness of the top left corner of the sample has been increased in Figure 8. The central region appeared porous as it readily absorbed the coloured polishing fluid. The mounted and polished disc B is also shown in Figure 8. It was much darker and did not show a colour gradient. Figure 8 also illustrates SEM images of the two polished discs. There was a clear difference in grain size and disc B evidenced grain pullout.

B)



Figure 7: A) the fractured surfaces of A) disc A made by SPS Condition A and B) disc B made by SPS condition B using commercial CeO₂.



Figure 8: An optical image and SEM image of the microstructure of A) disc A and B) disc B, respectively. The top left corner of disc A is shown with an increased image contrast and brightness to aid the identification of the border region. The green and yellow colouration is an effect caused by absorption of the polishing fluid.

Relatively intact pellets were made by spark plasma sintering lath-shaped CeO₂ powders 1 and 2 (see Table 1) using sintering condition C (see Figure 2). These will be referred to as SPS pellets C1 and C2, respectively (C refers to sintering condition C and the numbers refer to the CeO_2 batch numbers). These structures had uneven surfaces particularly near the edges as shown in Figure 9. In addition, the edges showed signs of fragmentation. These were all observed on removal from the SPS die except for the large exposed area shown in pellet C2 that was produced by a spontaneous fracture a week after SPS. The pellets were dark grey immediately after sintering but gradually changed back to a beige colour. This is shown in the fragmented regions in Figure 9. This occurred despite efforts to store the samples in thermally sealed bags with the air expelled, which were packed in another bag with desiccant inside to minimise water absorption. The relative densities of pellets C1 and C2 were 85 ± 0.5 % and 84 ± 0.5 %, respectively. Approximately 3.5 months after sintering, the pellets were removed from their vacuum packs. The samples fragmented and those associated with pellet C2 are shown in Figure 10. The fragments showed signs of colour change to beige. A significant portion of the SPS pellet C2 had turned beige with the central region still dark grey as illustrated in Figure 10. The pellet could not be sectioned lengthways or in cross-section due to insufficient handling strength. A fragment of the original pellet was cold mounted in epoxy resin, ground and polished (see Figure 11). As the sample was ground, more of the beige material was removed and dark grey regions became more visible with some evidence of cracking.



Figure 9: Images of the pellets (a) C1 and (b) C2 (nominally Ø13 mm x 16 mm) 1 week after SPS.



Figure 10: Two major fragments (left) of SPS pellet C2 approximately 3.5 months after manufacture, where the smaller fragment further broke up to a leave smaller piece (right, shown at higher magnification).



Figure 11: The polished surface of SPS pellet C2.

3.1.2 Vickers Hardness of SPS CeO₂

For disc A, ten indents that were made using a Mitutoyo MVK-G1 Hardness Tester were analysed using SEM and all were acceptable on the basis of morphology alone (per ASTM C1327-08 [31]). Indent dimensions were measured twice per indent to account for operator variability, which resulted in two sets of hardness measurements for the sample. The mean Vickers hardness with its standard deviation is shown in Table 3. The mean hardness values were very similar for the two measurement sets (2.7 GPa *cf* 2.8 GPa), which are combined into a single mean in Table 3.

On disc B, indent morphology was affected by the extensive pull-out on the surface caused during grinding and polishing, as shown in Figure 8. This pull-out was consistent with Figure 7, which showed the grains in the fractured surface were not well consolidated. There is an element of subjectivity in assessing the acceptability of indents. In this case, they were judged not to be acceptable, and so no Vickers hardness values are reported for disc B in Table 3.

The near-net shaped pellet C2, which was manufactured using lath-shaped CeO₂, did not yield a sufficient number of acceptable indentations to allow a hardness value to be determined. The indentations had large amounts of grain displacement around them. The low resistance to indentation of this material was not unexpected given that the pellet had fragmented prior to cold mounting.

Table 3	3: Com	parison	of	Vickers	hardness	measurements.
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Material	Sintering Technique/ Profile	Disc	Vickers Hardness (GPa)	Standard Deviation (GPa)
CeO ₂	Spark Plasma Sintering/ Condition A; see Figure 2.	A	2.7	0.32
CeO ₂	Cold-Pressing-and- Sintering; see Figure 3.	1	5.7	0.40
		2	5.5	0.26
		3	5.7	0.20
		5	6.4	0.25
Nd ₂ O ₃	1350 °C/50 MPa/3 min	20 mm disc	1.08	0.01

3.1.3 Cold-Press-and-Sinter Behaviour of Lath-shaped CeO₂

The powder particle properties of the input CeO_2 impacted the green disc handling strength to the extent that some powders could not be pressed successfully (batches 4 and 6 described in Table 1). Batches 1, 2, 3 and 5 were cold-pressed-and-sintered successfully and the dimensions and geometric densities are given in Table 4. The results show that the particle properties impact the sintered disc relative density. Discs 2 and 5 delaminated into three and two layers, respectively.

Batch #	Green Disc Thickness (mm)	Sintered Thickness (mm)	Sintered Diameter (mm)	Mass (g)	% Theoretical Density
1	2.21	2.10	6.78	0.4144	77 ± 10
2	2.01	1.90	6.99	0.4403	85 ± 12
3	1.80	1.70	7.39	0.4363	83 ± 12
5	1.83	1.80	7.01	0.3890	79 ± 12

Table 4: Successful cold-press-and-sinter CeO₂ trials

3.1.4 Vickers Hardness of Lath-shaped CeO₂

The mounted and polished intact discs are shown in Figure 12. Vickers hardness tests were carried out using a Buehler 1600-6300 Vickers hardness tester in the numbered regions shown in Figure 12.

SEM analysis was not required as the indents could be analysed optically with the indents in focus. The dark areas were regions infused with resin where specimen cracks were present. The average Vickers hardness values are outlined in Table 3.

The mean hardness values in Table 3 indicate that disc 5 was harder than disc 1, 2 and 3. The particle properties of the input material affected the Vickers hardness of the sintered body. Student's t-tests were conducted to separately compare the Vickers hardness data for discs 5 and 1, 5 and 2, and, finally, discs 5 and 3. Each of the three tests rejected the null hypothesis that mean Vickers hardness values of the two discs were the same to a confidence of greater than 95 %. Thus, the difference in the means are statistically significantly different in each case.



Figure 12: The polished surfaces of discs 1, 2, 3 and 5 cold-pressed-and-sintered discs (1 cm scale bars). Vickers hardness tests were conducted in the numbered regions.

3.2 Neodymium (III) Oxide

3.2.1 Demonstrating SPS of a Near-Net Shaped Pellet

Figure 13 illustrates the pellet sintered using commercial Nd_2O_3 and sintering profile C. It was intact and had a relative density of 98 %. Figure 13 shows that like the CeO_2 pellets, the SPS of Nd_2O_3 did not result in perfect cylinders; they had material absent near the circular faces.



Figure 13: The SPS Nd_2O_3 pellet (nominally Ø13 mm x 16 mm) sintered using Condition C (see Figure 2).

3.2.2 Sintering to a Target Relative Density Range

Figure 14 illustrates the relative densities of the discs with 20 mm nominal diameter sintered at different temperatures. To achieve a target relative density of 85-90 %, a near-net shaped pellet was sintered at 1080 °C. This pellet suffered some local edge fragmentation in a similar matter to the CeO₂ pellets manufactured by SPS as shown in Figure 9. The density of the resulting Nd₂O₃ pellet is also shown in Figure 14, and has a notably higher relative density (94 %) than predicted by the trend of the curve produced by sintering 20 mm diameters discs.



Figure 14: Illustrates the relative density versus sintering temperature results for the \emptyset 20 mm Nd₂O₃ discs (blue crosses) and for the near-net shaped pellet (orange rectangle).

3.2.3 Vickers Hardness of Nd₂O₃

Vickers hardness tests were conducted on the 1350 °C disc from Figure 14. This disc was selected in an attempt to generate a bounding maximum Vickers hardness (it was sintered to the highest density) value for Nd_2O_3 and to maximise the chance of generating acceptable indents. Unlike with the spark plasma sintered CeO_2 , the acceptability of the Vickers indents could be made with the optical microscope of the hardness test machine. Ten out of eleven indents were deemed acceptable. The Vickers hardness is reported in Table 3, which shows that the value was notably lower than CeO_2 .

4 Discussion

4.1 CeO₂

4.1.1 Spark Plasma Sintering Commercially Sourced and Lath-shaped CeO₂

Spark plasma sintering with graphite is known to provide a reducing environment. Nearly all the spark plasma sintered CeO_2 discs or pellets were grey, which was consistent with the observations made by Bevan [32] for CeO_{2-X} . Knachel et al. [33] investigated the reduction behaviour of CeO_2 and

produced samples with differing shades of grey or blue. They found that direct physical contact with graphite was not needed to cause the reduction. Thus, according to these pieces of literature, the grey appearance of the spark plasma sintered discs and pellets was indicative of the partial reduction of CeO₂. An example of sample outgassing during SPS was shown in Figure 6 for a disc fabricated using a one-stage process. The only sources for gas production would have been oxygen from the sample or the carbon monoxide or dioxide as a consequence of the reaction with the graphite die. It is likely that reduction of the CeO₂ as described by Zhou et al. [17] occurred during the experiment. Zhou et al. [17] attributed the observed weight loss in their CeO₂ powders to the reduction of the source material to Ce₂O₃. They state that the loss was particularly large beyond 1200 °C. The data presented in Figure 6 are in agreement.

Although an increase in temperature resulted in an increase in density for one-stage SPS, it was difficult to produce intact discs. A small change in SPS parameters resulted in a large change in disc density and in fracture surface microstructure. This was evidenced by both the one-stage and two-stage sintering trials. CeO₂ densification is thus difficult to control when using SPS. The 2-stage sintering profiles (condition A and B) were thought to have successfully produced intact 20 mm discs by permitting outgassing prior to densification. Although disc A was intact, the relative density was too low for the sintering conditions to be used further. The breakage in disc B whilst being removed from the die highlights that the handling strength of the material produced is still a concern.

On initial inspection, relatively intact pellets with relative densities in the mid-80 % were made using SPS when the particle properties of the input CeO_2 was changed from sub 5-µm agglomerates to larger lath-shaped particles, which were produced using the wet chemical process proposed for fuel manufacture. Both pellets C1 and C2 were not perfectly cylindrical due to edge fragmentation. Thus SPS could not achieve the desired geometry and appears unlikely to be able to do so with reproducibility.

The colour change in the SPS pellets (sintered using synthesised lath-shaped CeO₂) to dark grey and back to beige suggested reduction and re-oxidation, respectively. The suspected changes in oxygen to metal ratio and the irregular edges of the compacts would have added some uncertainty to the density measurements. The fragmentation following storage suggested the handling strength had diminished significantly. It is likely this oxidation is a factor in the fragmentation observed. Given that AmO₂ has a greater tendency to reduce [34], the complexity in spark plasma sintering AmO₂ is likely to be even greater.

4.1.2 Cold-Pressing-and-Sintering Lath-shaped CeO₂ with Varying Particle Sizes and Specific Surfaces Areas

Some of the cold-pressed-and-sintered ceria discs suffered from unwanted delamination. Americium oxide particles made via the oxalate precipitation and calcination route have similar lath morphologies; if the CeO₂ cold-press-and-sinter behaviour is indicative of AmO₂, then methods into preventing delamination may need to be investigated.

The physical characteristics of the CeO₂ starting powder plays an important role in being able to press integral discs. Batch 3 pressed particularly well. However, the inability to press batch 4 and yet to be able to press batch 1 is unexpected as they had similar specific surface areas and particle size distributions (apart from d_{90} values) according to Table 1. The green strengths of the intact discs were not measured but their integrities were qualitatively assessed as adequate. Discs 2 had a relative densities that met the target of 85-90 %, but disc 5 had a lower density (see Table 4). The

Vickers hardness of disc 5 was greater than that of disc 2. Table 1 shows that the powder in batch 5 had a similar specific surface area to batch 2, but the estimated particle sizes were notably lower in batch 5. These data show no evidence of the nature of the relationship between particle size, specific surface area, sintering behaviour and hardness. They do indicate that changes to the production parameters of the powder have a significant impact on final properties for this material. From this, it must be assumed that the situation will be similarly complex for americium oxide compounds.

4.1.3 Vickers Hardness Comparison of Spark Plasma Sintered and Cold-Pressed-and-Sintered CeO₂

Vickers hardness does not provide a direct indication of mechanical properties other than resistance to indentation. However, it provides an initial indication and means of comparison of the mechanical properties of these sintered CeO₂ discs and pellets. The processes necessary to prepare the specimen and generate valid hardness data do give an indication of potential practical issues with handling the material. Vickers hardness tests of the 2-stage spark plasma sintered discs A and B and pellet C2 indicated generally low resistance to indentation, with acceptable idents only achieved in disc A. The inability to obtain valid hardness data from an SPS pellet produced using lath-shaped CeO₂ particles indicates a potential concern for the use of this material (and AmO₂) and identifies an area for future work. Disc A had a much lower Vickers hardness than the cold-pressed-and-sintered discs (see Table 3). This again suggests inferior mechanical performance, perhaps attributable to the greater reduction during the sintering process disrupting the microstructure. A comparison of the measured Vickers hardness data in this study cannot be made with AmO_2 or Am_2O_3 due to the lack of published data. The mechanical properties of UO_2 are used as a reference for a nuclear material. The Vickers hardness of disc A was much lower than that of UO_2 (5.88 GPa) [35], whereas the coldpressed-and-sintered discs and pellets had hardness values that were consistent with UO2. This indicated their resistance to indentation was comparable to a nuclear material.

4.2 Spark Plasma Sintering Nd₂O₃

4.2.1 Initial Pellet Fabrication

It was demonstrated that a high relative density and intact Nd_2O_3 pellet can be produced by SPS. This was thought to be because Nd_2O_3 does not reduce as readily as CeO_2 as it is already in a lower oxidation state [36]. The sintering condition profile C (see § 3.2) resulted in a 98 % relative density. However, an Am_2O_3 pellet may not have the capacity to allow helium outgassing if Am_2O_3 were to behave like Nd_2O_3 under SPS i.e. to sinter to such a high density.

4.2.2 Demonstrating the Target Relative Density

Figure 14 illustrates that 20 mm diameter discs can be fabricated with relative densities of over 95 % at temperatures as low as 1200 °C. The disc sintered at 1350 °C had a comparable relative density to the pellet sintered using sintering profile C (Figure 2) despite the difference in aspect ratio and pressure. Although the near-net shaped pellet sintered at 1080 °C was expected to have a relative density within the target range (85-90 %) according to Figure 14, its relative density was much higher than expected at 94 \pm 0.2 %. The change in aspect ratio may have impacted the densification process. Future investigations need to be conducted to establish how target densities can be achieved for different ratios, particularly for a near-net shaped pellet geometry. More generally, the data present a clear drive for future studies of the sintering of these materials to use realistic aspect ratios in spite of the large volumes of material required.

The Vickers hardness value was much lower than the sintered CeO₂ (SPS and cold-pressed-andsintered) discs for which acceptable data could be collected (see Table 3). Thus ceria can be spark plasma sintered or cold-pressed-and-sintered to have superior hardness to spark plasma sintered Nd₂O₃. This is the first report of the hardness of Nd₂O₃ in the literature. Vickers hardness in ceramics is not a direct measurement of mechanical properties, and so the data only support the conclusion that there may be a concern about whether the handling integrity of Nd₂O₃ pellets is sufficient to make it an appropriate mechanical surrogate for americium oxide fuels.

5 Conclusions

Both the spark plasma sintering and cold-press-and-sinter techniques were able to sinter micrometric CeO₂ discs or pellets with relative densities within the target range of 85-90 % without ball-milling the surrogate fuel. Despite this, it has been shown that SPS is an unsuitable method to sinter CeO₂ due to the effects of reduction of the material. Intact high density discs could not be produced using commercial material and although relatively intact pellets could be made using larger lath-shaped particles (on initial inspection), the pellets exhibited chipping (initial evidence of fragmentation) and eventual further fragmentation. The study suggests that SPS cannot enable pellets to be made with reproducible geometries without further development. Further to this, as the CeO₂ pellets eventually disintegrated into fragments, which indicated that the current method cannot make pellets with stable integrity. Although this may have been caused by gradual oxidation, and the effect of annealing pellets post-SPS could be explored, this will not address the non-reproducible geometry. The design of pellet containment layers in the RPS system will require the reduction extent, crystallographic phase changes and thermal expansion to be characterised. Future X-ray diffraction studies will be conducted to assess the CeO₂ reduction behaviour.

Although cold-press-and-sinter takes longer than SPS it has enabled intact CeO_2 discs to be made with near-target densities and with higher Vickers hardness. All cold-pressed-and-sintered discs had comparable Vickers hardness to current nuclear ceramics (represented by uranium dioxide, UO_2), whereas the only meaningful value for the SPS discs was much lower. However, discs manufactured from lath-shaped material by cold-press-and-sinter did exhibit delamination. Furthermore, the effects of differences in particle size and specific surface area of the raw material were found to be significant without it being possible to correlate these effects in this study. Future studies must therefore fully characterise the effects of differences in lath-shaped particle size and specific surface area with the aim of creating a reproducible and well-characterised surrogate fuel form. It is important to note that AmO₂ produced via an oxalate precipitation and calcination route has similar lath-shaped morphology and the same considerations may apply.

High-density intact Nd_2O_3 discs were successfully produced via SPS, along with the capability to tailor the density in the required range by adjusting the sintering parameters. Using a modified sintering profile for a near-net shape pellet yielded an average density of 94 ± 0.2 % with some surface fragmentation. This illustrates that the aspect ratio of the pellet does affect the sintering parameters and behaviour. Future work on these materials must focus on representative geometries. Neodymium (III) oxide may be a suitable mechanical surrogate for Am_2O_3 ; however, the mean Vickers hardness, which was obtained from a limited data set, was much lower than for typical nuclear ceramics. This work requires extension to consider the use of cold-press and sinter, and the effect of raw powder obtained from a wet chemical process.

Cerium and neodymium oxides represent potential mechanical surrogates for the americium oxide fuel under development for the ESA space radioisotope power systems. On the basis of the current

studies, cold-press-and-sinter has fewer potential complications with reduction. Future research must consider the full effect of the wet-chemical processing parameters on particle characteristics and sintering behaviour in realistic pellet geometries.

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